

Stereospecific Stille-Coupling Reaction of (Z)-1,2-Bis(trimethylstannyl)ethenes with Hypervalent Iodonium Salts

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Abstract

The Stille coupling of (Z)-1,2-bis(trimethylstannyl)ethenes with hypervalent iodonium salts (1 equiv) proceeded stereospecifically in the presence of $PdCl_2$ (5 mol %) in DMF at room temperature to afford partially substituted (Z)-vinylstannanes under mild conditions. Alternatively, the use of 2 equivalents of hypervalent iodonium salts afforded tri-substituted alkenes. © 1999 Published by Elsevier Science Ltd. All rights reserved.

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The palladium-catalyzed cross-coupling of organostannanes with vinyl or aromatic halides and triflates has become an extremely powerful tool for the construction of carbon-carbon bonds [1]. This coupling reaction has been widely applied in organic synthesis due to the stability of organostannanes and tolerating many functional groups. However, the cross-coupling of bi-metallic reagents such as bisorganostannanes has received little attention. The stereospecific use of 1,1-bis(tributylstannyl)ethene in the Stille coupling has been recently reported by Quayle et al. [2]. The (Z)-1,2-bis(trimethylstannyl)ethene was coupled with bisiodo acetylide to construct enediyne moiety [3]. To the best of our knowledge no systematic and stereospecific coupling of (Z)-1,2-bis-stannanes was known. We assumed that the stereospecific coupling of 1,2-bis-stannanes in a sequential manner provides access to configurationally defined trisubstituted alkenes. In connection with our programs to utilize hypervalent iodonium salts in palladium-catalyzed cross-coupling with organostannanes [4], we have investigated the stereospecific coupling of (Z)-1,2-bis(trimethylstannyl)ethenes [5], which is shown in Scheme 1.

$$R^{1} + R^{2}I^{+}Ph BF_{4}$$

$$R^{2}I^{+}Ph BF_{4}$$

$$R^{2}I^{+}Ph$$

The stereospecific palladium-catalyzed cross-coupling of (Z)-1,2-bis(trimethylstannyl)-ethenes with hypervalent iodonium tetrafluoroborates is summarized in Table 1. The bis-

stannane 1a reacted with diphenyliodonium tetrafluoroborate (2a) in the presence of PdCl₂ (5 mol %) at room temperature for 30 min to afford the partially substituted stannane 3a [6] as the sole product in 80% yield (entry 1). As a control experiment the use of iodobenzene instead of diphenyliodonium salt 2a with bis-stannane 1a in the presence of PdCl₂ (5 mol %) at 80 °C for 10 h gave a complex mixture of products without giving the coupled product 3a. Treatment of 1a with p-methoxyphenyl(phenyl)iodonium tetrafluoroborate (2b) gave p-methoxyphenyl-substituted stannane 3b stereospecifically (entry 2). When the bis-stannane 1a was coupled with 2 equivalents of 2a, the trisubstituted olefin 4a was provided in 68% yield (entry 3). Alternatively, the hydroxymethyl-substituted bis-stannane 1b reacted with 1 equivalent of 2a and 2b to afford monosubstituted stannanes 3c [7] and 3d in 85 and 72% yields, respectively (entries 4 and 5). Treatment of 2a (1 equiv) with the carboethoxy-substituted bis-stannane 1c furnished the coupled stannane 3e in 83% yield (entry 7). For 1c, reaction with 2b and 2 equivalents of 2a gave the coupled products 3f and 4c in 81 and 64% yields, respectively (entries 8 and 9). Finally, the bis-stannane 1c was readily coupled with alkenyliodonium salt 2c (2 equiv) to afford dialkenyl-substituted ester 4d in 49% yield (entry 10).

Table 1. The Palladium-Catalyzed Cross-Coupling of 1,2-Bis(trimethylstannyl)ethenes with Hypervalent Iodonium Salts.

Entry	Substrate	Iodonium Salt	Product	Isolated Yield (%)
1	Ph Me ₃ Sn SnMe ₃	Ph ₂ I ⁺ BF ₄ - (1 equiv)	Ph SnMe ₃	80
	la la	2a	3a	
2	1a	p-MeOC ₆ H ₄ I ⁺ Ph BF ₄ (1 equiv)	p-MeOC ₆ H ₄ Ph SnMe	72
		2b	3b	
3	1a	Ph ₂ I ⁺ BF ₄ ⁻ (2 equiv)	Ph Ph	68
		2a	4a	
4	Me ₃ Sn SnMe ₃	2a (1 equiv)	Ph SnMe ₃	85
	1 b			
5	1 b	2b (1 equiv)	p-MeOC ₆ H ₄ SnMe	77
			3d	

The typical procedure is as follows. To a stirred solution of $Ph_2l^*BF_4^-$ (2a, 70.0 mg, 0.19 mmol) and $PdCl_2$ (1.1 mg, 5 mol %) in DMF (3 mL) at room temperature was added 1a (81.7 mg, 0.19 mmol). The reaction mixture was stirred for 30 min at room temperature. The mixture was filtered and evaporated in vacuo. The crude product was separated by SiO_2 column chromatography (hexanes, $R_f = 0.61$) to afford 3a (52.1 mg, 80%).

² To a stirred solution of 2a (140 mg, 0.39 mmol) and PdCl₂ (1.1 mg, 5 mol %) in DMF (5 mL) at room temperature was added 1a (81.7 mg, 0.19 mmol). The reaction mixture was stirred for 2 h at room temperature. The mixture was extracted with ether and the organic layer was dried over anhydrous MgSO₄ and evaporated *in vacuo*. The crude product was separated by SiO₂ column chromatography (hexanes, R_r = 0.43) to afford 4a (33.1 mg, 68%).

It is noteworthy that the stereospecific coupling can be achieved utilizing CuBr (5 mol %) as catalyst instead of using PdCl₂ in DMF at room temperature, which is shown in Scheme 2. The bis-stannanes 1a and 1c were coupled with 2a to give the partially substituted stannanes 3a and 3e in 75 and 72% yields (Scheme 2).

Scheme 2

Finally, this stereospecific coupling was applied to carbonylative coupling using carbon monoxide under atmospheric pressure in the presence of PdCl₂ (5 mol %) in DMF / DME (1:1) at room temperature for 30 min (Scheme 3). Accordingly the bis-stannane 1c was treated with 2a (1 equiv) to give the ketoolefin 3g resulting from destannylation during work-up in 78% yield. When the same reaction was conducted with 2 equivalents of 2a, diketoolefin 4e was obtained in 63% yield (Scheme 3).

In conclusion, the stereospecific Stille coupling of 1,2-bis(trimethylstannyl)ethenes with hypervalent iodonium salts was accomplished under extremely mild conditions.

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Satisfactory physical and spectral data were obtained in accord with the structure. Selected physical and spectra data as follow. 3a: 1H NMR (400 MHz, CDCl₃) 8 7.44 (s, 1H), 7.34~7.20 (m, 10H), 0.15~0.18 (m, 9H). IR (KBr) 3050, 1655, 1594 cm⁻¹. MS (EI): m/e (relative intensity) 342 (M⁺, 0.5), 327 (51), 178 (100), 163 (11). 3b: ¹H NMR (400 MHz, CDCl₃) δ 7.38~6.84 (m, 10H), 3.82 (S, 3H), 0.12~-0.11 (m, 9H). IR (KBr) 2950, 1600, 1249 cm⁻¹. MS (EI): m/e (relative intensity) 373 (M*, 5), 358 (100), 209 (94), 178 (31). 3c: ¹H NMR (400 MHz, CDCI₃) 8 7.40 (s, 1H), 7.30~7.19 (m, 5H), 4.42 (m, 2H), 0.11~0.03 (m, 9H), IR (KBr) 3398, 1654, 1443, 1067 cm⁻¹. MS (EI): m/e (relative intensity) 296 (M⁻, 2), 281 (21), 165 (100). 163 (80), 133 (37). 3d: ¹H NMR (400 MHz, CDCl₃) δ 7.32 (s, 1H), 7.13 (m, 2H), 6.85 (m, 2H), 4.39 (m, 2H), 3.81 (s, 3H), 0.22~0.09 (m, 9H). IR (KBr) 3400, 1607, 1508, 1247 cm⁻¹. MS (EI): m/e (relative intensity) 326 (M⁻, 1), 296 (8), 166 (100), 162 (61). 3e: H NMR (400 MHz, CDCl₃) δ 8.37 (s, 1H), 7.35~7.26 (m, 5H), 4.25 (q, 2H, J = 7.0 Hz), 1.34 (t, 3H, J = 7.0 Hz), 0.14~0.02 (m. 9H). IR (KBr) 1702, 1635, 1213 cm⁻¹. MS (EI): m/e (relative intensity) 339 (M⁻, 0.01), 294 (25), 146 (100), 131 (23), 108 (18). 3f: H NMR (400 MHz, CDCl₃) δ 8.30 (s, 1H), 7.24~7.20 (m, 2H), 6.87~6.84 (m, 2H), 4.21 (q, 2H, J = 7.0 Hz), 3.81 (s, 3H), 1.29 (t, 3H), 3H, J = 7.0 Hz), 0.19~-0.02 (m, 9H). IR (KBr) 1700, 1605, 1250, 1171 cm⁻¹. MS (EI): m/e (relative intensity) 369 (M⁺, 0.25), 354 (70), 310 (92), 161 (100). 3g: ¹H NMR (400 MHz, CDCl₃) δ 8.02 (m, 2H), 7.99 (d, 1H, J = 15.0 Hz), 7.62 (m, 1H), 7.53 (m, 2H), 6.88 (d, 1H, J = 15.0 Hz), 4.30 (q, 2H, J = 7.0 Hz), 1.35 (t, 3H, J = 7.0 Hz). IR (KBr) 1720, 1672, 1240 cm⁻¹. MS (EI): m/e (relative intensity) 204 (M $^{\circ}$, 11), 159 (15), 130 (24), 105 (100), 77 (44). **4d**: ¹H NMR (400 MHz, CDCl₃) δ 7.54 \sim 6.90 (m, 15H), 4.32 (q, 2H, J = 7.0 Hz), 1.39 (t, 3H, J = 7.0 Hz). IR (KBr) 1705, 1655, 1622, 1232 cm⁻¹. MS (El): m/e (relative intensity) 304 (M⁻, 67), 258 (70), 230 (100), 154 (16), 77 (19). 4e: ¹H NMR (400 MHz, CDCl₃) δ 8.15 (s, 1H), 7.97~7.89 (m, 4H), 7.63~7.44 (m, 6H), 4.28 (q, 2H, J = 7.0 Hz). 1.22 (t, 3H, J = 7.0 Hz). IR (KBr) 1722, 1669, 1263 cm⁻¹. MS (EI): m/e (relative intensity) 308 (M*, 1), 165 (100), 77 (33).

Reference and Notes

- [1] For reviews, see: (a) Stille JK. Angew. Chem. Int. Ed. Engl. 1986;25:508-524. (b) Mitchell TN. Synthesis 1992;803-815. (c) Farina, V. In Comprehensive Organometallic Chemistry II; Abel EW, Stone FGA, Wilkinson G, Ed. Pergamon Press: New York, 1995; Vol. 12, Chapter 3,4; pp 200-221. (d) Farina V. Pure Appl. Chem. 1996;68:73-78.
- [2] Quayle P, Wang J, Xu J, Urd CJ. Tetrahedron Lett. 1998;39:485-488.
- [3] (Z)-1,2-bis(stannyl)ethenes: (a) Shair MD, Yoon T, Danishefsky, SJ. J. Org. Chem. 1994:59:3755-3757. (b) Mitchell TN, Kwetkat K, Rutschow D, Schnelder U. Tetrahedron 1989:45:969-978. (E)-1,2-bis(stannyl)ethenes: (c) Catellani M. Luzzati S, Musco A, Speroni F. Synthetic Metals 1994:62:223-228. (d) Djuric SW, Haack RA, Yu SS. J. Chem. Soc. Perkin Trans. 1 1989:2133-2134. (e) Haack RA, Penning TD, Djuric SW, Dziuba JA. Tetrahedron Lett. 1988:29:2783-2786.
- [4] (a) Kang SK, Lee HW, Kim JS, Choi SC. Tetrahedron Lett. 1996;37:3723-3725. (b) Kang SK, Lee HW. Jang SB, Kim TH, Kim JS. Synth. Commun. 1996;26:4311-4318.
- [5] (a) Piers E, Skerlj RT. Can. J. Chem. 1994;72:2468-2482. (b) Mitchell TN, Amamria A, Killing H, Rutschow DJ. Organomet. Chem. 1986;304:257-265.
- [6] The compound 3a was treated with MeOH/AcOH (10:1) at room temperature to give trans-stilbene in 80% yield.
- [7] The (Z)-stereochemistry was confirmed by NOE-experiment.